



## Piceanonols A, and B, triterpenoids bearing a novel skeletal system isolated from the bark of *Picea jezoensis* var. *hondoensis*

Reiko Tanaka,<sup>a,\*</sup> Tsuyoshi Kumagai,<sup>a</sup> Yasuko In,<sup>a</sup> Toshimasa Ishida,<sup>a</sup> Hoyoku Nishino<sup>b</sup> and Shunyo Matsunaga<sup>a</sup>

<sup>a</sup>Department of Medicinal Chemistry, Osaka University of Pharmaceutical Sciences, 4-20-1 Nasahara, Takatsuki, Osaka 569-1094, Japan

<sup>b</sup>Kyoto Prefectural University of Medicine, Kamigyo-ku, Kyoto 602-0841, Japan

Received 22 April 1999; accepted 9 June 1999

## **Abstract**

A CHCl<sub>3</sub> extract of the stem bark of *Picea jezoensis* var. *hondoensis* (Pinaceae) afforded two triterpenoids named piceanonols A and B with a novel  $14(13\rightarrow12)abeo-12\alpha H$ -serratane skeleton. Extensive NMR studies, MS spectral analyses and X-ray crystallographics established their structures as  $3\beta$ -methoxy- $14\beta$ ,21 $\beta$ -dihydroxy- $14(13\rightarrow12)abeo-12\alpha H$ -serratan-13-one and  $21\alpha$ -methoxy- $14\beta$ -hydroxy- $14(13\rightarrow12)abeo-12\alpha H$ -serratane-3,13-dione. © 1999 Elsevier Science Ltd. All rights reserved.

From our series of studies on biologically active constituents from Pinaceae plants, we previously reported the isolation of five new serratane triterpenoids from the bark of *Picea jezoensis* var. hondoensis.<sup>1-3</sup> Continuous fractionation of the CHCl<sub>3</sub> extract of the bark further afforded two novel triterpenoids named piceanonols A (1) and B (2).

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Corresponding author.

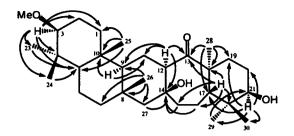


Figure 1. Selected HMBC correlations for 1

Piceanonol A (1)<sup>5</sup> was shown to have the molecular formula  $C_{31}H_{52}O_4$  by the high resolution MS spectrum. The IR spectrum showed hydroxyl absorption ( $\nu_{max}$  3540 cm<sup>-1</sup>) and ketone ( $\nu_{max}$  1690 cm<sup>-1</sup>). The <sup>1</sup>H and <sup>13</sup>C NMR spectra showed seven tertiary methyl groups, 10 methylenes, four methines of which a methine group appeared in the low magnetic field [ $\delta_H$  2.99 (1H, dd, J=12.3, 3.4 Hz)], four quaternary carbons, a methine group attached to a secondary methoxy group [ $\delta_H$  2.62 (1H, dd, J=11.2, 4.5 Hz), 3.36 (3H, s);  $\delta_C$  57.5 (q), 88.6 (d)], a methine group attached to a secondary hydroxyl group [ $\delta_H$  3.35 (1H, t, J=2.4 Hz);  $\delta_C$  75.6 (d)], a tertiary hydroxyl group [ $\delta_C$  71.8 (s)] and a ketone [ $\delta_C$  217.7 (s)]. Compound 1 was therefore a pentacyclic triterpene ketone with a secondary and a tertiary alcohol, and a secondary methoxy group. However, the <sup>1</sup>H and <sup>13</sup>C chemical shift values of 1 were considerably different from those of serratanes.<sup>1-4</sup>

In the EIMS spectrum, 1 showed fragment ion peaks at m/z 347.2586 (ion a, 17%), m/z 334.2504 (ion b, 100%), m/z 319.2288 (b-Me, 46%), m/z 153.1269 (ion c, 22%) and m/z 135.1176 (c-H<sub>2</sub>O, 76%) and these peaks are characteristic of the cleavage of the D ring.

The acetylation of 1 gave a monoacetate (1a),  $^6$   $v_{max}$  3521 (tert OH), and its hydroxymethine proton signal was shifted to  $\delta$  4.66. Treatment of 1 with boiling dry THF containing 0.5% H<sub>2</sub>SO<sub>4</sub> gave conjugated  $\Delta^{12(14)}$ -en-13-one (1b) [ $\delta_C$  132.3 (s), 136.5 (s)] and  $\Delta^{14}$ -en-13-one (1c) [ $\delta_H$  4.99 (1H, d, J=2.5 Hz);  $\delta_C$  131.4 (s), 133.7 (d)]. From these spectral data and chemical conversion, 1 was assumed to have a novel carbon skeleton and revealed the presence of O=C-CH-C-OH moiety in the molecule.

The selected HMBC correlations for 1 were shown in Fig. 1. The NOESY spectrum of 1 showed cross-peaks between signals of Me-25 with Me-24 and Me-26, Me-28 with H-12 $\alpha$  and Me-29, and H-12 $\alpha$  with H-9 $\alpha$ . All these data indicated that 1 has a novel  $14(13 \rightarrow 12)abeo$ -12 $\alpha$ H-serratane having a ketone and a tertiary hydroxyl group at the C-13 and C-14 positions.

The absolute structure was finally established by the X-ray analysis of its p-bromobenzoate (1d).<sup>8</sup> Fig. 2 shows the ORTEP view of 1d, and 1 possesses a novel 6-6-6-7-6 ring system, which has not been described in the literature.

Piceanonol B (2)<sup>7</sup> was shown to have the molecular formula  $C_{31}H_{52}O_4$  by the high-resolution MS spectrum. The IR spectrum showed hydroxyl absorption ( $\nu_{max}$  3550 cm<sup>-1</sup>) and ketone ( $\nu_{max}$  1705 and 1698 cm<sup>-1</sup>). The <sup>1</sup>H and <sup>13</sup>C NMR spectra resembled those of 1 except for the presence of a ketone [δ<sub>C</sub> 217.4 (s)] instead of a secondary hydroxyl group. The EIMS spectrum of 2 showed fragment ion peaks, corresponding to those of 1, at m/z 331.1992 (ion a, 68%), m/z 318.2185 (ion b, 100%), m/z 303.1992 (b-Me, 59%), m/z 167 (ion c, 22%) and m/z 135.1158 (c-MeOH, 63%). From the above data, compound 2 was assumed to be 21α-methoxy-14β-hydroxy-14(13 $\rightarrow$ 12)abeo-12αH-serratane-3,13-dione, and extensive 2D NMR studies including HMBC and NOESY spectra supported this structure. To confirm this, we conducted the X-ray analysis of 2. Fig. 2 shows the ORTEP view of 2,8 and accordingly compound 2 had the same 14(13 $\rightarrow$ 12)abeo-12αH-serratane skeleton of 1 with a tertiary hydroxyl group at C-14β, an equatorial methoxy group at C-21, and a ketone at C-3.

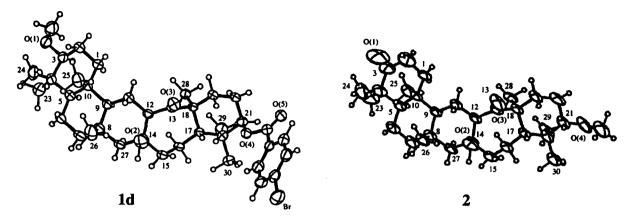


Figure 2. The ORTEP view of piceanonol A p-bromobenzoate (1d) and piceanonol B (2)

Plausible biogenetical routes of these compounds from  $\Delta^{13}$ -serratenes are shown in Scheme 1. These compounds are particularly interesting since this is the first report of *abeo*-serratane carbon skeletons.

Scheme 1. Plausible biogenetical routes of compounds 1 and 2

In the anti-tumor-promoting assay (Table 1),<sup>9</sup> compound 2 exhibited 100 and 20.4% inhibition in the 100 and 25  $\mu$ g/ml concentrations, respectively, against the 12-*O*-tetradecanoylphorbol-13-acetate (TPA)-enhanced incorporation of <sup>32</sup>Pi into phospholipids of cultured HeLa cells. Although no conspicuous

Table 1
Inhibitory effects of 1, 1b, 1c and 2 on TPA-enhanced <sup>32</sup>Pi-incorporation into phospholipids of HeLa cells

Inhibition %				
Compounds	100 μg/ml	50 μg/ml	25 μg/ml	10 μg/mi
1	0	0	0	0
1b	-	-	2.6	-
1c	-	-	10.6	-
2	100	26.8	20.4	14.5

inhibitory effects were observed for 1 (0%) in the 25  $\mu$ g/ml concentration, 1b and 1c, the dehydration products of 1 increased the effects up to 2.6 and 10.6%, respectively.

## Acknowledgements

The authors are grateful to Mrs. Michiko Yoneda and Mrs. Mihoyo Fujitake of this University for NMR and MS measurements.

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- 5. Compound 1, prisms, mp 241–243°C (MeOH–CHCl<sub>3</sub>);  $[\alpha]_D^{23}$  +7.6 (c 0.80, CHCl<sub>3</sub>); IR (KBr)  $\nu_{max}$  cm<sup>-1</sup>: 3540 (OH), 2939, 2874, 1690 (C=O), 1463, 1388, 1106; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  0.76 (s, Me-24), 0.84 (s, Me-25), 0.87 (s, Me-29), 0.95 (s, Me-23), 0.98 (s, Me-30), 1.13 (s, Me-26), 1.24 (s, Me-28), 2.62 (1H, dd, J=11.2, 4.5 Hz, H-3 $\alpha$ ), 2.99 (1H, dd, J=12.3, 3.4 Hz, H-12 $\alpha$ ), 3.36 (s, MeO), and 3.35 (1H, t, J=2.4 Hz, H-21 $\alpha$ ); EIMS: m/z 488.3883 (M<sup>+</sup>, 2%, C<sub>31</sub>H<sub>52</sub>O<sub>4</sub> requires 488.3865), 470 (17%), 347 (17%), 334 (100%), 319 (46%), 221 (6%), 189 (19%), 153 (22%), 135 (76%).
- 6. Compound **1a**, mp 250–253°C (MeOH–CHCl<sub>3</sub>),  $[\alpha]_D^{23}$  +34.0 (*c* 0.46, CHCl<sub>3</sub>); IR (KBr)  $v_{max}$  cm<sup>-1</sup>: 3521 (OH), 1741 and 1242 (OAc), 1705 (C=O); M<sup>+</sup>: m/z 530.3968 (C<sub>33</sub>H<sub>56</sub>O<sub>5</sub>, requires 530.3968). Compound **1b**,  $\delta_C$  132.3 (s, C-12), 136.5 (s, C-14); M<sup>+</sup>: m/z 470. Compound **1c**,  $\delta_H$  4.99 (1H, d, J=2.5 Hz, H-12 $\alpha$ );  $\delta_C$  131.4 (s, C-14), 133.7 (d, C-15); M<sup>+</sup>: m/z 470. Compound **1d**, mp >300°C (MeOH–CHCl<sub>3</sub>); M<sup>+</sup>: m/z 671.
- 7. Compound **2**, prisms, mp 266–268°C (MeOH–CHCl<sub>3</sub>);  $[\alpha]_D^{23}$  +34.0 (*c* 0.46, CHCl<sub>3</sub>); IR (KBr)  $\nu_{max}$  cm<sup>-1</sup>: 3550 (OH), 2938, 2871, 1705 and 1698 (C=O), 1457, 1418 (CH<sub>2</sub>C=O), 1387, 1102; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  0.79 (s, Me-29), 0.95 (s, Me-25), 0.99 (s, Me-30), 1.04 (s, Me-24), 1.08 (s, Me-23), 1.17 (s, Me-26), 1.23 (s, Me-28), 2.66 (1H, dd, J=12.2, 4.8 Hz, H-21 $\beta$ ), 3.04 (1H, dd, J=12.3, 3.5 Hz, H-12 $\alpha$ ) and 3.36 (s, MeO); EIMS: m/z 486.3715 (M<sup>+</sup>, 2%, C<sub>31</sub>H<sub>50</sub>O<sub>4</sub> requires 486.3706), 331 (68%), 318 (100%), 303 (59%), 286 (16%), 205 (9%), 189 (6%), 167 (22%), 135 (63%).
- 8. Crystal data of 1d and 2. 1d: C<sub>38</sub>H<sub>55</sub>O<sub>5</sub>Br, M=671.757, monoclinic, space group P2<sub>1</sub>, a=7.225 (3), b=37.268 (12), c=7.011 (1) Å; β=111.94 (2); U=1750.9 (10) ų, Dc=1.2741 g cm<sup>-3</sup>, Z=2. 2: C<sub>31</sub>H<sub>52</sub>O<sub>4</sub>, M=486.737, monoclinic, space group P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, a=17.146 (3), b=21.610 (7), c=7.393 (1) Å; V=2739.3 (10) ų, Dc=1.180 g cm<sup>-3</sup>, Z=4. A single crystal was used for X-ray diffraction data collection on a Rigaku AFC-5 diffractometer employing graphite-monochromated CuKα radiation. A total of 3047 (1d) and 2682 (2) independent reflections within 2θ=130° were collected in an ω-2θ scan mode and were corrected for the Lorenz and polarization factors. The structures were solved by direct methods and refined by least-squares analysis with use of the anisotropic temperature factors for non-H-atoms and isotropic factors for H-atoms, where H-atoms were observed on a final difference Fourier map. The discrepancy indices R and R<sub>w</sub> are 0.0568 and 0.0707 for 2978 [F<sub>0</sub>>0.0] reflections for 1d, and 0.0757 and 0.0502 for 2549 [F<sub>0</sub>>0.0] reflections for 2, respectively. The atomic coordinates, anisotropic thermal parameters, and bond lengths and angles have been deposited at the Cambridge Crystallographic Data Center. See Notice to Authors, Issue No. 1.
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